

Cytotoxicity evaluation of calcium hypochlorite and other commonly used root canal irrigants against human gingival fibroblast cells: An in vitro evaluation

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Abstract

Background. The conventional endodontic therapy primarily focuses on biomechanical preparation, which is achieved by the application of various intracanal irrigants and intracanal medicaments. One of the most commonly used intracanal irrigants – sodium hypochlorite (NaOCl) – has already been proven to have an antimicrobial effect as well as the ability to dissolve tissues in the areas where files cannot reach. One of the recently used irrigants having a promising effect is calcium hypochlorite (Ca(OCl)₂), which has been shown to be relatively more stable than NaOCl and has much more chlorine ions.

Objectives. The aim of this study was to assess the individual cytotoxicity of various root canal irrigants and the combined cytotoxicity of NaOCl and Ca(OCl)₂ with ethylenediaminetetraacetic acid (EDTA) against human gingival fibroblast (hGF) cells.

Material and methods. The evaluation of the individual cytotoxicity was carried out with regard to the following root canal irrigants: NaOCl; Ca(OCl)₂; and chlorhexidine (CHX). The evaluation of the combined cytotoxicity regarded NaOCl/EDTA and Ca(OCl)₂/EDTA. The concentrations used were 0.025%, 0.050%, 0.10%, and 0.20%. The cytotoxicity against hGF cells was examined within a timeframe of 6 h and 24 h with the use of the sulforhodamine B (SRB) assay.

Results. It was observed that Ca(OCl)₂ had a mean absorbance rate of 0.315 ± 0.02 , 0.294 ± 0.03 , 0.265 ± 0.03 , and 0.240 ± 0.02 at 0.025%, 0.050%, 0.10%, and 0.20%, respectively. In combination with EDTA, the mean absorbance rate was 70.12 ± 2.9 , 67.42 ± 4.3 , 64.35 ± 3.6 , and 61.58 ± 4.1 at 0.025%, 0.050%, 0.10%, and 0.20%, respectively. The cytotoxic effect of the root canal irrigants on hGF cells was observed to be statistically significant ($p < 0.05$).

Conclusions. Calcium hypochlorite is less cytotoxic than NaOCl, and when used in combination with EDTA, it was shown to have its cytotoxic effect on hGF cells reduced to a great extent.

Key words: chlorhexidine, EDTA, endodontics, root canal irrigants, sodium hypochlorite

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Introduction

The typical aim in the endodontic therapy is to perform the optimal disinfection of the root canal system, after which a suitable filling material is placed in this disinfected space, preventing any further ingress of microorganisms leading to further complications.¹ It is well known that microorganisms are a critical factor in the development of pulp and periapical diseases, and have an influence on the prognosis of the endodontic therapy in the long run.^{2,3} Currently, along with mechanical instrumentation, various adjuvants, such as intracanal irrigants and intracanal medicaments are used, which helps to significantly reduce the amount of microorganisms. Due to the complexity of the root canal anatomy, intracanal irrigants are of great importance in the endodontic procedure, since they have the ability to get into the sites where mechanical instrumentation fails to reach.⁴

Various intracanal irrigants have been used in the conventional endodontic therapy in the past decades in order to help the clinician to completely eradicate microbes, necrotic tissues and debris from the root canal system, the most common being chlorhexidine (CHX) and sodium hypochlorite (NaOCl), which have been shown to have an antimicrobial effect, the ability to inactivate endotoxins as well as the ability to remove some part of the smear layer and necrotic tissue remnants.⁵ Sodium hypochlorite, one of the most common intracanal irrigants, also shows antimicrobial activity and is able to dissolve pulpal tissue remnants in spaces such as fins and ramifications, where instrumentation cannot reach.⁶ The capability of NaOCl to dissolve pulpal tissue results from its alkaline nature as well as from the release of HOCl^- and OCl^- ions.⁷

Sodium hypochlorite is known to show certain drawbacks, such as the lack of substantivity, the ability to dissolve only the organic portions of the smear layer and a cytotoxic effect when extruded into the periapical region. One of the ways to overcome these drawbacks is to use it in combination with ethylenediaminetetraacetic acid (EDTA), which reduces the pH of NaOCl in a time-dependent manner as well as the amount of the available free chlorine, making NaOCl more stable and preventing the weakening of the dentinal matrix.⁸ One of the recently introduced intracanal irrigants of a promising potential is calcium hypochlorite ($\text{Ca}(\text{OCl})_2$), which has shown an antimicrobial effect and at the same time a much higher chlorine release in comparison with NaOCl.⁹ These irrigants have been reported to affect various cells present in the periapical complex. Although numerous assessments of their cytotoxicity against multiple cell lines have been performed, there is no data available in the dental literature on the cytotoxic effect of these irrigants on human gingival fibroblast (hGF) cells. The current study aimed to evaluate the individual cytotoxic activity of various root canal irrigants, and the combined cytotoxicity of NaOCl and $\text{Ca}(\text{OCl})_2$ with EDTA against hGF cells.

Material and methods

This in vitro study had received prior approval of the institutional review board (No. SRB/SDC/ENDO-1805/20/04).

Preparation of the cell line

This in vitro study was conducted under aseptic conditions in a research laboratory. The hGF cell lines were received from the National Centre for Cell Science, Pune, India. The hGF cells were seeded in the α -minimal essential medium (α -MEM) containing 10% fetal bovine serum (FBS), 100 IU/mL penicillin, 2.5 $\mu\text{g}/\text{mL}$ streptomycin, 2.5 $\mu\text{g}/\text{mL}$ amphotericin B, and 50 $\mu\text{g}/\text{mL}$ ascorbic acid, which was replaced twice a week, and incubated in 5% CO_2 at 37°C for humidification.

Preparation of calcium hypochlorite and other irrigants

The CaOH_2 powder of 65% purity (Merck, Darmstadt, Germany) was weighed on a precision balance and mixed with distilled and sterilized water. When the solutions were completely dissolved, they were filtered twice to remove debris, and then stored in bottles. Other chemicals used in the study were as follows: NaOCl (Molychem, Mumbai, India); EDTA (Thermo Fisher Scientific, Waltham, USA); and CHX (Sigma-Aldrich, St. Louis, USA).

Experimental groups

The individual cytotoxicity evaluation was done in 4 groups:

- group A: 10% NaOCl diluted to 0.025%, 0.050%, 0.10%, and 0.20%;
- group B: 10% $\text{Ca}(\text{OCl})_2$ diluted to 0.025%, 0.050%, 0.10%, and 0.20%;
- group C: 2% CHX diluted to 0.025%, 0.050%, 0.10%, and 0.20%;
- group D: the growth medium used as a control group.

For the combined cytotoxicity evaluation, 2 groups were analyzed:

- group A: 10% NaOCl diluted to 0.025%, 0.050%, 0.10%, and 0.20%, mixed with EDTA;
- group B: 10% $\text{Ca}(\text{OCl})_2$ diluted to 0.025%, 0.050%, 0.10%, and 0.20%, mixed with EDTA.

Cytotoxicity test with the sulforhodamine B assay

The cytotoxicity test was performed using a modification of the method developed by Vichai and Kirtikara.¹⁰ The assessment was carried out within a time-frame of 6 h and 24 h, with each concentration being tested in triplicate. The microtiter plates were incubated with Dulbecco's Modified Eagle's Medium (DMEM)

(Life Technologies, Grand Island, USA) supplemented with 10% FBS, 100 μL /well of the culture medium, for 24 h at 37°C with 5% CO_2 . After 24 h of incubation, cell adhesion and growth were observed. The cellular concentration used was 1.4×10^4 cells/mL. The medium was then discarded, and the treatment group samples, controls and DMEM without FBS were added to each well. After 24 h of incubation, the cells were fixed with trichloroacetic acid (TCA) (100 μL /well) and placed under refrigeration for 1 h at 4°C. Trichloroacetic acid was then removed, and the plates were washed in low-flow water 3 times and dried. After that, the plates were stained for 20 min at room temperature with sulforhodamine B (SRB) (Sigma-Aldrich) 0.4% (50 μL /well). The plates were washed with 1% acetic acid solution and dried at room temperature. The unbound dye was removed after washing; the dye bound to the protein was solubilized in the basic medium to determine the optical density in a plate reader set at 570 nm. The colorimetric evaluation gives an estimate of total protein mass. The cellular concentration used was 1.4×10^4 cells/mL. The percentage of living cells for each concentration of the tested substances was assessed with the following equation (Equation 1):

$$\text{percentage of cells survived} = \frac{(\text{AbsT} - \text{AbsC})}{(\text{AbsC})} \times 100 \quad [\%] \quad (1)$$

where:

AbsT – average absorbance of the tested substances;

AbsC – average absorbance of the control.

Statistical analysis

The statistical analysis was carried out using IBM SPSS for Windows, v. 21.0 (IBM Corp., Armonk, USA). The one-way analysis of variance (ANOVA) was applied to assess differences in absorbance at 570 nm at different time intervals between the treatment groups, post-hoc Tukey's test was used for the intergroup comparison of absorbance between the test groups; the independent *t* test was done to assess cell viability at different time points between the groups. The level of significance was set at $p < 0.05$.

Results

The experiments were carried out in triplicate. The results of the assessment of cytotoxicity against hGF cells are shown in a dose-dependent and time-dependent manner, with a total timeframe of 6 h and 24 h. They are presented as mean (*M*) \pm standard deviation (*SD*).

Table 1 shows the absorbance values for different root canal irrigants at different concentrations at 6 h and 24 h. Table 2 presents the percentage of cell viability for different root canal irrigants at different concentrations at 6 h and 24 h. The findings of the study suggest that CHX, NaOCl and $\text{Ca}(\text{OCl})_2$ had an effect on hGF cells ($p = 0.001$). Figure 1 shows the dose-dependent effect of various root canal irrigants on hGF cells at 6 h. The intergroup analysis of the rate of absorbance at the 0.025% concentration showed statistically significant differences in various combinations: NaOCl vs $\text{Ca}(\text{OCl})_2$ ($p = 0.002$); NaOCl vs negative control (NC) ($p = 0.001$); $\text{Ca}(\text{OCl})_2$ vs CHX ($p = 0.013$); and CHX vs NC ($p = 0.001$).

Table 1. Absorbance values for various root canal irrigants at different concentrations at 6 h and 24 h

Irrigant	Absorbance at 570 nm at 6 h				Absorbance at 570 nm at 24 h			
	0.025%	0.050%	0.10%	0.20%	0.025%	0.050%	0.10%	0.20%
NaOCl	0.125 \pm 0.04	0.129 \pm 0.01	0.105 \pm 0.01	0.052 \pm 0.01	0.286 \pm 0.02	0.258 \pm 0.01	0.222 \pm 0.05	0.202 \pm 0.04
$\text{Ca}(\text{OCl})_2$	0.315 \pm 0.02	0.294 \pm 0.03	0.265 \pm 0.03	0.240 \pm 0.02	0.498 \pm 0.06	0.468 \pm 0.02	0.424 \pm 0.03	0.390 \pm 0.03
CHX	0.175 \pm 0.03	0.148 \pm 0.02	0.112 \pm 0.1	0.098 \pm 0.02	0.312 \pm 0.09	0.238 \pm 0.04	0.221 \pm 0.06	0.192 \pm 0.05
NC	0.392 \pm 0.04	0.388 \pm 0.03	0.381 \pm 0.04	0.396 \pm 0.05	0.512 \pm 0.03	0.507 \pm 0.03	0.512 \pm 0.04	0.512 \pm 0.05

NaOCl – sodium hypochlorite; $\text{Ca}(\text{OCl})_2$ – calcium hypochlorite; CHX – chlorhexidine; NC – negative control.
Data presented as mean (*M*) \pm standard deviation (*SD*).

Table 2. Cell viability for various root canal irrigants at different concentrations at 6 h and 24 h

Irrigant	Percentage of viable cells [%] at 6 h				Percentage of viable cells [%] at 24 h			
	0.025%	0.050%	0.10%	0.20%	0.025%	0.050%	0.10%	0.20%
NaOCl	52.78 \pm 2.6	41.89 \pm 1.6	32.60 \pm 2.5	20.80 \pm 1.4	51.78 \pm 4.3	48.56 \pm 2.6	39.89 \pm 2.8	23.82 \pm 2.1
$\text{Ca}(\text{OCl})_2$	69.89 \pm 1.8	60.23 \pm 3.4	53.99 \pm 1.6	50.16 \pm 2.4	56.64 \pm 5.6	50.11 \pm 0.6	42.71 \pm 3.1	28.15 \pm 1.9
CHX	79.35 \pm 3.2	71.54 \pm 2.4	68.21 \pm 3.8	70.25 \pm 3.9	70.25 \pm 6.4	63.58 \pm 3.1	59.96 \pm 4.6	53.89 \pm 4.3
NC	100.00 \pm 9.6	100.00 \pm 8.2	100.00 \pm 7.5	100.00 \pm 5.7	100.00 \pm 8.1	100.00 \pm 2.6	100.00 \pm 8.3	100.00 \pm 9.1

Data presented as mean (*M*) \pm standard deviation (*SD*).

The comparison of absorbance at the 0.050% concentration showed statistically significant differences in the following cases: NaOCl vs $\text{Ca}(\text{OCl})_2$ ($p = 0.001$); NaOCl vs NC ($p = 0.001$); $\text{Ca}(\text{OCl})_2$ vs CHX ($p = 0.001$); $\text{Ca}(\text{OCl})_2$ vs NC ($p = 0.001$); and CHX vs NC ($p = 0.001$). At the 0.10% concentration, differences in absorbance were observed to be statistically significant for NaOCl vs $\text{Ca}(\text{OCl})_2$ ($p = 0.001$), NaOCl vs NC ($p = 0.001$), $\text{Ca}(\text{OCl})_2$ vs CHX ($p = 0.001$), $\text{Ca}(\text{OCl})_2$ vs NC ($p = 0.002$), and CHX vs NC ($p = 0.001$). At the 0.20% concentration, the comparison of absorbance showed statistically significant differences for NaOCl vs $\text{Ca}(\text{OCl})_2$ ($p = 0.001$), NaOCl vs NC ($p = 0.001$), $\text{Ca}(\text{OCl})_2$ vs CHX ($p = 0.005$), $\text{Ca}(\text{OCl})_2$ vs NC ($p = 0.006$), and CHX vs NC ($p = 0.001$).

Figure 2 shows the dose-dependent effect of various root canal irrigants on hGF cells at 24 h. The comparison of the absorbance of various irrigants at 0.025% showed statistically significant differences for the following: NaOCl vs $\text{Ca}(\text{OCl})_2$ ($p = 0.004$); NaOCl vs NC

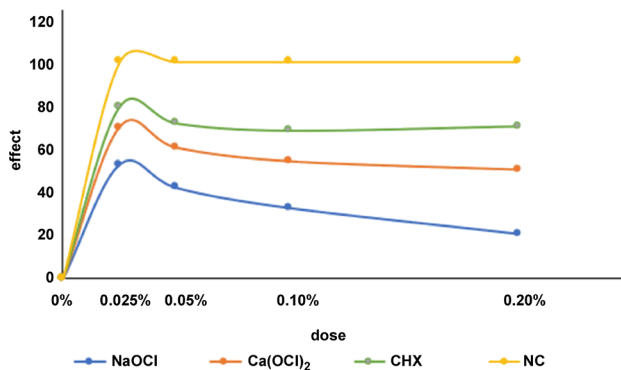


Fig. 1. Dose-effect plot of sodium hypochlorite (NaOCl), calcium hypochlorite ($\text{Ca}(\text{OCl})_2$) and chlorhexidine (CHX) at 6 h against human gingival fibroblast (hGF) cells (sulforhodamine B (SRB) assay)

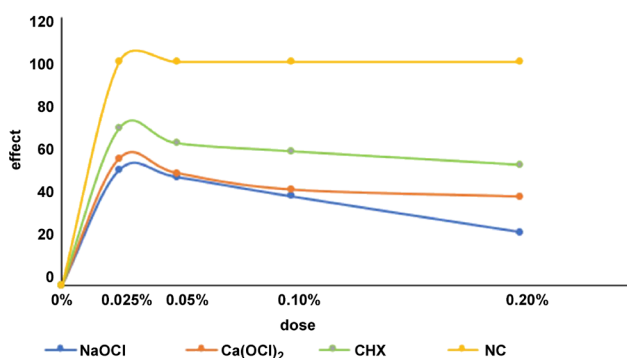


Fig. 2. Dose-effect plot of sodium hypochlorite (NaOCl), calcium hypochlorite ($\text{Ca}(\text{OCl})_2$) and chlorhexidine (CHX) at 24 h against human gingival fibroblast (hGF) cells (sulforhodamine B (SRB) assay)

($p = 0.003$); $\text{Ca}(\text{OCl})_2$ vs CHX ($p = 0.004$), $\text{Ca}(\text{OCl})_2$ vs NC ($p = 0.003$), $\text{Ca}(\text{OCl})_2$ vs NC ($p = 0.008$), and CHX vs NC ($p = 0.005$). At the 0.050% concentration, the comparison of absorbance showed statistically significant differences for NaOCl vs $\text{Ca}(\text{OCl})_2$ ($p = 0.002$), NaOCl vs NC ($p = 0.001$), $\text{Ca}(\text{OCl})_2$ vs CHX ($p = 0.001$), and CHX vs NC ($p = 0.001$). At the 0.10% concentration, the intergroup comparison of absorbance showed statistically significant differences for NaOCl vs $\text{Ca}(\text{OCl})_2$ ($p = 0.001$), NaOCl vs NC ($p = 0.001$), $\text{Ca}(\text{OCl})_2$ vs CHX ($p = 0.001$), and CHX vs NC ($p = 0.001$). In the case of the 0.20% concentration, the intergroup absorbance rate was observed to be statistically significantly different for NaOCl vs $\text{Ca}(\text{OCl})_2$ ($p = 0.008$), NaOCl vs NC ($p = 0.001$), $\text{Ca}(\text{OCl})_2$ vs CHX ($p = 0.008$), $\text{Ca}(\text{OCl})_2$ vs NC ($p = 0.006$), and CHX vs NC ($p = 0.001$).

Table 3 shows the percentage of cell viability for hGF cells post reaction with NaOCl/EDTA and $\text{Ca}(\text{OCl})_2$ /EDTA at different concentrations at 6 h and 24 h. Table 4 presents the final concentrations of NaOCl and $\text{Ca}(\text{OCl})_2$ after serial dilution with EDTA at 6 h and 24 h. Figure 3 depicts the combined cytotoxicity against hGF cells at 6 h of the combinations of NaOCl/EDTA and $\text{Ca}(\text{OCl})_2$ /EDTA. The comparison of cell viability done at 6 h showed statistically significant difference between NaOCl/EDTA and $\text{Ca}(\text{OCl})_2$ /EDTA at 0.025% ($p = 0.001$), 0.050% ($p = 0.001$), 0.10% ($p = 0.001$), and 0.20% ($p = 0.001$). Figure 4 shows the combined cytotoxicity against hGF cells at 24 h for the combinations of NaOCl/EDTA and $\text{Ca}(\text{OCl})_2$ /EDTA. The comparison of cell viability done at 24 h showed statistically significant difference between NaOCl/EDTA and $\text{Ca}(\text{OCl})_2$ /EDTA at 0.025% ($p = 0.019$), 0.050% ($p = 0.007$), 0.10% ($p = 0.005$), and 0.20% ($p = 0.029$).

Table 4. Final concentration after serial dilution with ethylenediaminetetraacetic acid (EDTA) at 6 h and 24 h

Time [h]	NaOCl/EDTA	$\text{Ca}(\text{OCl})_2$ /EDTA
6	0.025/0.1	0.005/0.2
	0.050/0.2	0.012/0.3
	0.100/0.3	0.025/0.4
	0.200/0.4	0.037/0.5
24	0.037/0.01	0.015/0.05
	0.015/0.02	0.037/0.04
	0.023/0.03	0.050/0.03
	0.041/0.04	0.075/0.02

Table 3. Cell viability at 6 h and 24 h in combination with ethylenediaminetetraacetic acid (EDTA) (sulforhodamine B (SRB) assay)

Combination of irrigant/EDTA	Percentage of viable cells [%] at 6 h				Percentage of viable cells [%] at 24 h			
	0.025%	0.050%	0.10%	0.20%	0.025%	0.050%	0.10%	0.20%
NaOCl/EDTA	51.24 \pm 3.1	50.43 \pm 2.3	39.43 \pm 3.1	30.56 \pm 1.9	50.12 \pm 2.1	41.89 \pm 3.5	34.16 \pm 2.4	28.49 \pm 2.9
$\text{Ca}(\text{OCl})_2$ /EDTA	70.86 \pm 4.5	66.89 \pm 5.8	64.32 \pm 2.8	63.47 \pm 2.8	70.12 \pm 2.9	67.42 \pm 4.3	64.35 \pm 3.6	61.58 \pm 4.1

Data presented as mean (M) \pm standard deviation (SD).

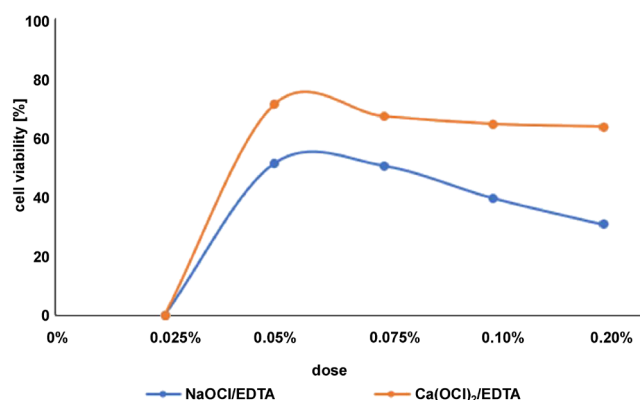


Fig. 3. Dose–cell viability plot of sodium hypochlorite (NaOCl) with ethylenediaminetetraacetic acid (EDTA) and calcium hypochlorite ($\text{Ca}(\text{OCl})_2$) with EDTA at 6 h against human gingival fibroblast (hGF) cells (sulforhodamine B (SRB) assay)

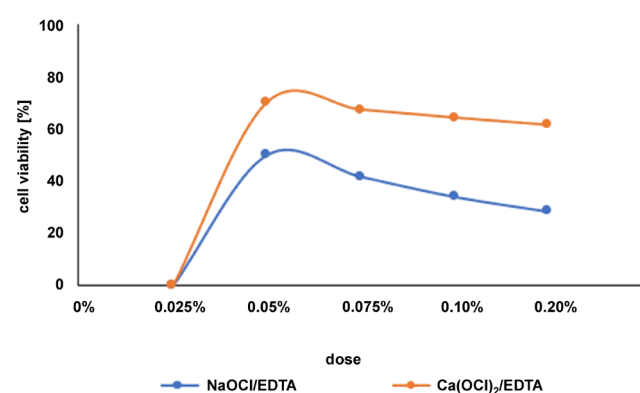


Fig. 4. Dose–cell viability plot plot of sodium hypochlorite (NaOCl) with ethylenediaminetetraacetic acid (EDTA) and calcium hypochlorite ($\text{Ca}(\text{OCl})_2$) with EDTA at 24 h against human gingival fibroblast (hGF) cells (sulforhodamine B (SRB) assay)

Discussion

Root canal irrigants have a pivotal role in the success of the endodontic therapy. It is well known that they participate in removing necrotic tissue and debris from fins and ramifications, where mechanical instrumentation cannot reach. Still, there is a risk that during the endodontic procedure, intracanal irrigants will be extruded periapically.¹¹ This can cause damage to cells and also impair wound healing.¹² Another potential effect of the extrusion of irrigants is that they may impair the survival of stem cells, which are required for successful tissue regeneration.¹³

Human gingival fibroblast cells are reported to have a much higher reparative potential than human periodontal ligament fibroblasts (hPDLF).¹⁴ Though hPDLF cells could simulate periodontal ligament fibroblasts better, they are known to produce a higher amount of collagen than hGF cells, and could potentially influence the results of cytotoxicity for the materials tested.¹⁵ Apart from having great regenerative properties as compared to other fibroblasts, hGF cells also have the multilineage capability to differentiate into different types of cells, such as osteoblasts and periodontal ligament cells, which can be potentially

used for periodontal tissue engineering.¹⁶ The presence of lipopolysaccharides in endodontic pathogens has an immunosuppressive effect on the human pulp, leading to pulpal pain and periapical inflammation.¹⁷ Unlike other cells present in the periodontium, hGF cells do not express lipopolysaccharide tolerance and have been shown to release inflammatory cytokines for up to 7 days, which makes them the least susceptible to microbial action.¹⁸

As compared to other standard cell lines, hGF cells are diploid host cells, continuous in nature, with the specialized properties retained; they are more likely to demonstrate the cytotoxic effect exerted by dental materials on cells than other commercially available cell lines, which are aneuploid in nature, showing a heteroploid chromosome pattern and responding differently to different materials.¹⁹ The use of commercially available standardized cell lines has an advantage that they are easily accessed, but can demonstrate greater toxicity levels and show an altered effect of the use of dental materials as compared to specific lineage cell lines, which are more desirable, as they bridge the gap between in vitro and in vivo biocompatibility.²⁰

Various studies have already been done in the field of endodontics, using hGF cells. Barnhart et al. assessed the cytotoxicity of various intracanal irrigants with the CyQUANT[®] assay and found potassium iodide and calcium hydroxide to be the least cytotoxic against hGF cells.²¹ Lee et al. also evaluated the cytotoxicity of various dental adhesives against hGF cells and found that all of the tested adhesives had a cytotoxic effect on hGF cells.²²

In the present study, the SRB assay was used for the cytotoxicity assessment, as it has been shown to have a higher sensitivity; the cell viability count is not affected by any compounds released from the specimen, is independent of any cellular metabolic action, which makes the assay much easier to conduct in comparison with other assay methods.²³

Calcium hypochlorite is another intracanal irrigant introduced in endodontics which has been shown to have a good antimicrobial activity.^{24–26} Though it is very similar to NaOCl in terms of chemical action, the presence of more chlorine (Cl^-) in $\text{Ca}(\text{OCl})_2$ results in more pronounced antimicrobial action and a greater tissue dissolving capability as compared to NaOCl.²⁷ The current study showed the least cytotoxic effect in the case of CHX. It is known that CHX has a sustained antimicrobial activity as well as low toxicity, but its major disadvantage is its inability to dissolve organic pulp tissues.²⁸

The present study showed that $\text{Ca}(\text{OCl})_2$ maintained the viability of cells when tested within a timeframe of 6 h and 24 h. In contrast, NaOCl was proven to decrease the levels of viable hGF cells after 24 h even further. In an in vitro study, Coaguila-Llerena et al. found that differences in cytotoxicity against L929 and hPDLF cells between $\text{Ca}(\text{OCl})_2$ and 2% CHX were statistically insignificant; also, the cellular proliferation capacity was maintained when $\text{Ca}(\text{OCl})_2$ at a concentration of 2.5% was used.²⁹ In another study,

Ferraz Blattes et al. conducted an in vitro cytotoxicity assessment against 3T3 fibroblasts and an analysis of the inflammatory reaction in rats; the authors reported that $\text{Ca}(\text{OCl})_2$ showed lower levels of inflammation, better healing response as well as lower cytotoxicity as compared to NaOCl.³⁰

The combined use of NaOCl with EDTA is the recommended protocol to be followed according to the American Association of Endodontists (AAE) guidelines and was the basis for evaluating the combined cytotoxic effect in our study. It is known that NaOCl/EDTA acts on the organic/inorganic portion of the smear layer present in dentin and shows a synergistic effect by chelating Ca^{2+} ions, causing collagen deproteinization, which results in a superficial change in dentin, preventing further microbial colonization.³¹ The present study also showed that cytotoxicity was reduced to a greater extent with the addition of EDTA. Our results are consistent with those obtained by Vouzara et al., who proved that the application of the combination of NaOCl and EDTA reduced the cytotoxic effect against MRC5 cells.³² The results obtained in our study could be due to an antagonizing effect of EDTA on NaOCl, which plays a critical role in reducing its pH in a time-dependent manner as well as decreasing exponentially the amount of the released chlorine gas.³³ Due to the chemically similar nature, the same action could be justified in the case of $\text{Ca}(\text{OCl})_2$ as well.

Limitations


One of the limitations of the present study is not using an in vivo model to assess the cytotoxic reaction, since an in vitro model is an ideally conditioned environment and may not necessarily transmit the achieved results in vivo. Another limitation is probably the use of lower concentrations of irrigants, which were diluted further to different levels for assessment standardization and may not necessarily show the same result when used in their clinically recommended concentrations.


Conclusions

Based on the present study, it can be concluded that $\text{Ca}(\text{OCl})_2$ showed less cytotoxicity in a time-dependent and dose-dependent manner than NaOCl, but was more cytotoxic against hGF cells in comparison with chlorhexidine gluconate. The combined use of $\text{Ca}(\text{OCl})_2$ and NaOCl with EDTA showed a more significant reduction of cytotoxicity against hGF cells. In contrast, the sustainability of hGF cells after 24 h was much greater in the case of using $\text{Ca}(\text{OCl})_2$ as compared to NaOCl.

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